

### Tutorial Sheet-3

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#### Example 3-1

In a steady-flow system, a substance flows at the rate of 4 kg/s. It enters at a pressure of  $620 \text{ kN/m}^2$ , a velocity of 300 m/s, internal energy 2100 kJ/kg, and specific volume  $0.37 \text{ m}^3/\text{kg}$ . It leaves the system at a pressure of  $130 \text{ kN/m}^2$ , a velocity of 150 m/s, internal energy 1500 kJ/kg and specific volume  $1.2 \text{ m}^3/\text{kg}$ . During its passage through the system the substance has a loss by heat transfer of  $30 \text{ kJ/kg}$  to the surroundings.

Determine the power of the system in kilowatts (kW), stating whether it is from or to the system. Neglect any change in potential energy.

#### Solution:

#### Steady-flow system (open system)

Given:

$$\dot{m} = 4 \text{ kg/s} \quad . \quad \text{mass flow rate}$$

$$P_1 = 620 \text{ kN/m}^2 \quad . \quad \text{inlet pressure}$$

$$P_2 = 130 \text{ kN/m}^2 \quad . \quad \text{outlet pressure}$$

$$C_1 = 300 \text{ m/s} \quad . \quad \text{inlet velocity}$$

$$C_2 = 150 \text{ m/s} \quad . \quad \text{outlet velocity}$$

$$u_1 = 2100 \text{ kJ/kg} \quad . \quad \text{internal energy at inlet}$$

$$u_2 = 1500 \text{ kJ/kg} \quad , \quad \text{internal energy at outlet}$$

$$v_1 = 0.37 \text{ m}^3/\text{kg} \quad , \quad \text{specific volume at inlet}$$

$$v_2 = 1.2 \text{ m}^3/\text{kg} \quad , \quad \text{specific volume at out let}$$

Heat lost,  $Q = 30 \text{ kJ/kg}$  (must be negative value)

Calculate power of system in (KW),  $W^\circ = ?$

Potential energy= neglected i.e  $g(Z_2 - Z_1) = 0$

General relation of Steady flow energy equation(SFEE) for unit mass:

$$Q - W = (h_2 - h_1) + \frac{C_2^2 - C_1^2}{2} + g(Z_2 - Z_1)$$

$$h_1 = u_1 + P_1 v_1 = 2100 + 620 \times 0.37 = 2329.4 \text{ kJ/kg}$$

$$h_2 = u_2 + P_2 v_2 = 1500 + 130 \times 1.2 = 1656 \text{ kJ/kg}$$

$$\frac{C_2^2}{2} = \frac{(150)^2}{2 \times 10^3} = 11.25 \text{ kJ/kg}$$

$$\frac{C_1^2}{2} = \frac{(300)^2}{2 \times 10^3} = 45 \text{ kJ/kg}$$

$$\frac{C_2^2 - C_1^2}{2} = 11.25 - 45 = -33.75 \text{ kJ/kg}$$

Substitute these values in general relation, we get

$$(-30) - W = (1656 - 2329.4) - 33.75$$

$$W = 677.15 \text{ kJ/kg}$$

$$\begin{aligned} \text{power of the system } W^\circ &= m^\circ \times W = 4 \times 677.15 \\ &= 2708.6 \text{ kJ/S or } 2708.6 \text{ kW} \end{aligned}$$

### Example 3-2

10 kg of fluid per minute goes through a reversible steady flow process. The properties of fluid at the inlet are :  $p_1=1.5 \text{ bar}$  ,  $v_1 = 0.038 \text{ m}^3/\text{kg}$  ,  $C_1=110 \text{ m/s}$  and  $u_1=910 \text{ kJ/kg}$  and at the exit are  $p_2=5.5 \text{ bar}$  ,  $v_2 = 0.181 \text{ m}^3/\text{kg}$  ,  $C_2=190 \text{ m/s}$  and  $u_2=710 \text{ kJ/kg}$ . During the passage , the fluid rejects  $55 \text{ kJ/s}$  and rises through 55 meters. Determine:

(i) the change in enthalpy ( $\Delta h$ )

(ii) Work done during the process (W).

**Solution.**

$$\text{Mass flow rate , } m^\circ = 10 \text{ kg/min} = \frac{10}{60} = 0.166 \text{ kg/s}$$

Properties of fluid at inlet :

Pressure  $p_1=1.5 \text{ bar}$

Specific volume  $v_1 = 0.038 \text{ m}^3/\text{kg}$

Velocity  $C_1=110 \text{ m/s}$

Internal energy  $u_1=910 \text{ kJ/kg}$

Properties of fluid at the exit :

Pressure  $p_2=5.5 \text{ bar}$

Specific volume  $v_2 = 0.181 \text{ m}^3/\text{kg}$

Velocity  $C_2=190 \text{ m/s}$

Internal energy  $u_2=710 \text{ kJ/kg}$

Heat rejected by the fluid ,  $Q=55 \text{ kJ/s}$

The fluid is rises through 55 m, i.e  $Z_2 = 55 \text{ m}$

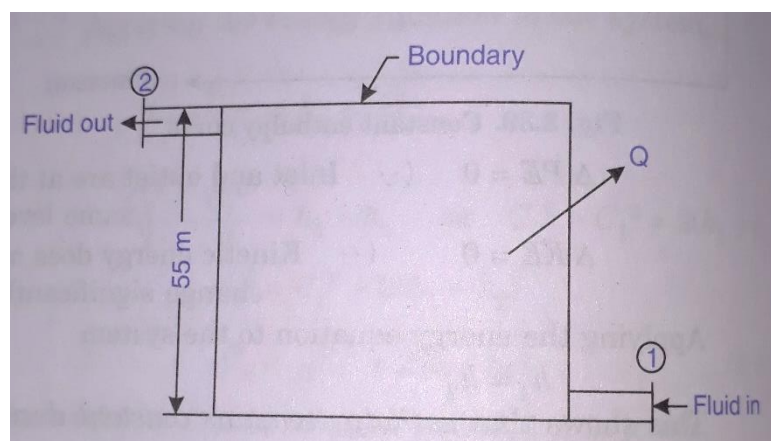
(i) The change in enthalpy,

$$\Delta h = \Delta u + \Delta(pv)$$

$$\begin{aligned}\Delta(pv) &= p_2v_2 - p_1v_1 = (5.5 \times 10^2) \times 0.181 - (1.5 \times 10^2) \times 0.038 \\ &= 99.55 - 5.7 = 93.85 \text{ kJ/kg}\end{aligned}$$

$$\Delta u = u_2 - u_1 = 710 - 910 = -200 \text{ kJ/kg}$$

$$\Delta h = \Delta u + \Delta(pv) = -200 + 93.85 = -106.15 \text{ kJ/kg} \quad (\text{Ans.})$$



(ii) The steady flow equation for unit mass flow can be written as

$$Q = \Delta KE + \Delta PE + \Delta h + W$$

Where Q is the heat transfer per kg of fluid

$$Q^\circ = Q \times m^\circ \quad \text{or} \quad Q = \frac{Q^\circ}{m^\circ} = \frac{55}{0.166} = 331.32 \text{ kJ/kg}$$

Now,

$$\Delta KE = \frac{c_2^2 - c_1^2}{2} = \frac{(190)^2 - (110)^2}{2} = 12000 \text{ Nm/kg or } 12000 \text{ J/kg or } 12 \text{ kJ/kg}$$

$$\Delta PE = (Z_2 - Z_1)g = (55 - 0)9.81 = 539.5 \text{ Nm or J} = 0.54 \text{ kJ/kg}$$

Substituting the value in steady flow equation,

$$-331.32 = 12 + 0.54 - 106.15 + W$$

$$W = -236.77 \text{ kJ/kg}$$

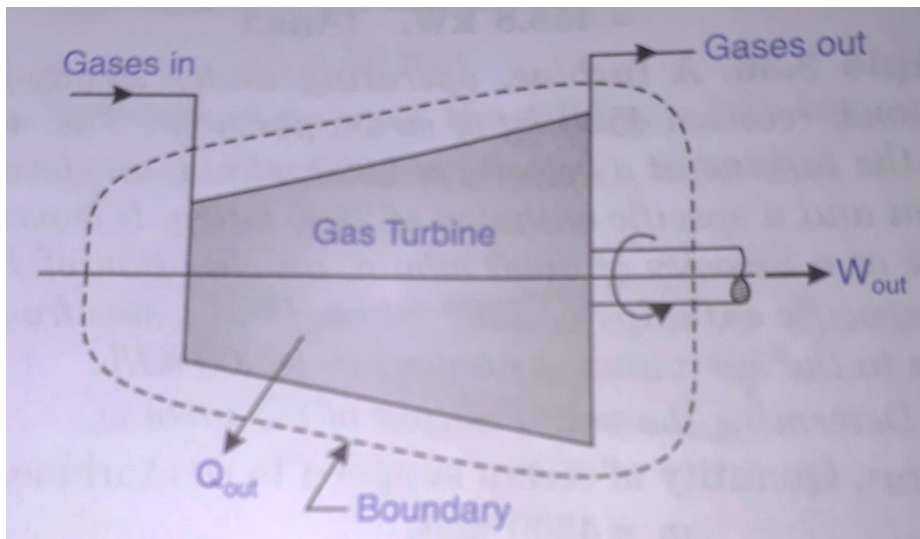
$$\text{Work done per second, } \dot{W} = \dot{m} \times W = 0.166 \times -236.77 = -39.30 \text{ kW} \quad (\text{Ans.})$$

### Example 3-3

In a gas turbine unit, the gases flow through the turbine is 15 kg/s and the power developed by the turbine is 12000 kW. The enthalpies of gases at the inlet and outlet are 1260 kJ/kg and 400 kJ/kg respectively, and the velocity of gases at the inlet and outlet are 50 m/s and 110 m/s respectively. Calculate :

(i) The rate at which heat is rejected from the turbine

(ii) The area of the inlet pipe given that the specific volume of the gases at the inlet is 0.45 m<sup>3</sup>/kg.



**Solution.**

Rate of flow of gases,  $\dot{m} = 15 \text{ kg/s}$

Specific volume at inlet  $v_1 = 0.45 \text{ m}^3/\text{kg}$

Power developed by turbine,  $\dot{W} = 12000 \text{ kW}$

Work done,  $W = \frac{\dot{W}}{\dot{m}} = \frac{12000}{15} = 800 \text{ kJ/kg}$

Enthalpy of gases at the inlet,  $h_1 = 1260 \text{ KJ/kg}$

Enthalpy of gases at the outlet,  $h_2 = 400 \text{ KJ/kg}$

Velocity of gases at the inlet,  $C_1 = 50 \text{ m/s}$

Velocity of gases at the outlet,  $C_2 = 110$  m/s

**(i) Heat rejected, Q**

Using the Steady Flow Energy Equation (S.F.E.E)

Neglecting the potential energy since  $Z_1 = Z_2$

$$h_1 + \frac{c_1^2}{2} + Q = h_2 + \frac{c_2^2}{2} + W \dots\dots\dots (1)$$

Kinetic energy at inlet

$$= \frac{C_1^2}{2} = \frac{50^2}{2 \times 1000} = 1.25 \text{ kJ/kg}$$

Kinetic energy at outlet

$$\frac{C_2^2}{2} = \frac{110^2}{2 \times 1000} = 6.05 \text{ kJ/kg}$$

Substituting these value in eqn. (1)

$$1260 + 1.25 + Q = 400 + 6.05 + 800$$

$$Q = -55.2 \text{ KJ/kg}$$

i.e, Heat rejected = +55.2 KJ/kg

$$= 55.2 \times 15 = 828 \text{ kW}$$

**(ii) Inlet area,  $A_1$**

Using the relation,

$$m^\circ = \frac{CA}{v}$$

$$\therefore A_1 = \frac{v_1 m^\circ}{C_1} = \frac{0.45 \times 15}{50} = 0.135 \text{ m}^2$$

$$A_1 = \frac{\pi d^2}{4}$$

$$0.135 = \frac{\pi d^2}{4}$$

$$d^2 = 0.135 \times \frac{4}{\pi} = 0.171$$

$$d = 0.413 \text{ m}$$